



Bachelor in Physics (Academic Year 2024-25)

Materials Physics			Code	800510	Year	3 rd	Sem.	1 st
Module	Applied Physics	Topic	Applied Physics-Compulsory		Character	Optative		

	Total	Theory	Exercises
ECTS Credits	6	4.2	1.8
Semester hours	45	31	14

Learning Objectives (according to the Degree's Verification Document)
<ul style="list-style-type: none"> • Know the structure and the main physical properties of materials. • Be able to recognize and establish the basic relationships between the microstructure and physical properties of materials. • Know the possibilities of controlling the properties of materials through their design. • Acquire the basic notions about the applications of different types of materials.
Brief description of contents
Crystals, disordered and amorphous solids; structure and physical properties of materials; alloys; preparation of materials; nanomaterials; materials in micro- and nanoelectronics; Ceramic materials.
Prerequisites

Coordinator	Paloma Fernández Sánchez			Dept.	FM
	Office	02.115.0	e-mail	arana@fis.ucm.es	

Theory/Problems – Schedule and Teaching Staff								
Group	Lecture Room	Day	Time	Professor	Period/Dates	Hours	T/E	Dept.
B	4A	T, Th	14:00 – 15:30	Leonor Chico Gómez	Full term	45	T/E	FM

T: Theory, E: Exercises

Office hours				
Group	Professor	Schedule	E-mail	Location
B	Leonor Chico Gómez	X: 11.00 -13.00 J: 16.00-17.00 +3h On line	leochico@ucm.es	02.122.0

Syllabus
<p>1. Crystals, disordered and amorphous solids. Short and long-range structural order. Single-, poly- and nanocrystalline materials. Crystalline materials: crystal systems and lattices. Cohesion: primary and secondary bonds. Micro- and nanostructures. Real crystals: defects, surface, diffusion processes.</p> <p>2. Structure and physical properties of materials. Relationship between structure and properties. Metals, ceramics, semiconductors, polymers and soft matter, composites. Preparation and design of materials. Phase transformations.</p> <p>3. Mechanical properties. Elasticity, anelasticity, plasticity. Hardening. Degradation mechanics. Properties at the nanoscale.</p> <p>4. Electrical properties. Electronic conduction: metals and semiconductors. Ionic conduction. Dielectrics (ferro- and piezoelectricity). Nanostructures and quantum confinement. Materials in micro- and nanoelectronics.</p> <p>5. Optical properties. Light absorption and emission. Photoconductivity. Nanostructures in optoelectronic devices.</p> <p>6. Magnetic properties. Origin of magnetism. Dia- and paramagnetism. Hard and soft magnets. Magnetic nanostructures.</p> <p>7. Thermal properties. Thermal expansion and conductivity. Thermoelectric effect, generation of heat and cooling.</p>

Bibliography
<p>Basic</p> <ul style="list-style-type: none"> - "Understanding solids. The Science of Materials". Richard Tilley, Wiley, 2004 - "The science and engineering of materials" D. Askeland, W. Wright. Cengage Learning has currently 4 similar titles available as both e-book paper (https://www.cengage.co.uk/search/?keyword=askeland) - "Materials Science and engineering", W. D. Callister and D.G. Rethwisch, Wiley, 2020 <p>Advanced</p> <ul style="list-style-type: none"> - "Introduction to Soft Matter", Ian W. Hamley, Wiley (2000) - "Nanomaterials: An Introduction to Synthesis, Properties and Applications", Dieter Vollath, Wiley, 2008
Online Resources
<p><i>Campus virtual and professor's web page pilot0.fis.ucm.es/paloma and links therein</i></p>

Methodology
<p>Lectures to explain the fundamental concepts that will include examples and applications. For these classes, computer projection will be used fundamentally. The students will have the material used in class in advance. Practical classes to solve exercises. The work performed during these sessions will be a part of the evaluation.</p>

Evaluation Criteria		
Exams	Weight:	70%
The exam will consist of a series of theoretical and practical questions (of a similar level to those solved		

in class).		
The use of books will not be allowed.		
Other Activities	Weight:	30%
For the evaluation the exercises carried out in class and participation in classes will be considered.		
Final Mark		
Final mark will be $N_{Final}=0.7N_{Exam}+0.3N_{OtherActiv}$, where N_{Exam} and $N_{OtherActiv}$ are (from 0 to 10) the marks obtained in the two previous sections.		
The mark in the extraordinary call of June/July will be obtained exactly with the same evaluation procedure.		